

FUNDAMENTAL PHYSICAL CONSTANTS

Peter J. Mohr and Barry N. Taylor

These tables give the 1998 self-consistent set of values of the basic constants and conversion factors of physics and chemistry recommended by the Committee on Data for Science and Technology (CODATA) for international use. The 1998 set replaces the previous set of constants recommended by CODATA in 1986; assigned uncertainties have been reduced by a factor of 1/5 to 1/12 (and sometimes even greater) relative to the 1986 uncertainties. The recommended set is based on a least-squares adjustment involving all of the relevant experimental and theoretical data available through December 31, 1998. Full details of the input data and the adjustment procedure are given in Reference 1.

The 1998 adjustment was carried out by P. J. Mohr and B. N. Taylor of the National Institute of Standards and Technology (NIST) under the auspices of the CODATA Task Group on Fundamental Constants. The Task Group was established in 1969 with the aim of periodically providing the scientific and technological communities with a self-consistent set of internationally recommended values of the fundamental physical constants based on all applicable information available at a given point in time. The first set was published in 1973 and was followed by a revised set first published in 1986; the current 1998 set first appeared in 1999. In the future, the CODATA Task Group plans to take advantage of the high level of automation developed for the current set in order to issue a new set of recommended values at least every four years.

At the time of completion of the 1998 adjustment, the membership of the Task Group was as follows:

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REFERENCES

1. Mohr, Peter J., and Taylor, Barry N., *J. Phys Chem. Ref. Data* **28**, 1713, 1999; *Rev. Mod. Phys.* **72**, 351, 2000. The 1998 set of recommended values is also available at the Web site of the Fundamental Constants Data Center of the NIST Physics Laboratory: <http://physics.nist.gov/constants>.

Fundamental Physical Constants

Quantity	Symbol	Value	Unit	Relative std. uncert. u_r
UNIVERSAL				
speed of light in vacuum	c, c_0	299 792 458	m s^{-1}	(exact)
magnetic constant	μ_0	$4\pi \times 10^{-7}$ $= 12.566\ 370\ 614\dots \times 10^{-7}$	N A^{-2} N A^{-2}	
electric constant $1/\mu_0 c^2$	ϵ_0	$8.854\ 187\ 817\dots \times 10^{-12}$	F m^{-1}	(exact)
characteristic impedance of vacuum $\sqrt{\mu_0/\epsilon_0} = \mu_0 c$	Z_0	376.730 313 461...	Ω	(exact)
Newtonian constant of gravitation	G	$6.673(10) \times 10^{-11}$	$\text{m}^3 \text{ kg}^{-1} \text{ s}^{-2}$	1.5×10^{-3}
	$G/\hbar c$	$6.707(10) \times 10^{-39}$	$(\text{GeV}/c^2)^{-2}$	1.5×10^{-3}
Planck constant in eV s	h	$6.626\ 068\ 76(52) \times 10^{-34}$ $4.135\ 667\ 27(16) \times 10^{-15}$	J s eV s	7.8×10^{-8} 3.9×10^{-8}
$h/2\pi$ in eV s	\hbar	$1.054\ 571\ 596(82) \times 10^{-34}$ $6.582\ 118\ 89(26) \times 10^{-16}$	J s eV s	7.8×10^{-8} 3.9×10^{-8}
Planck mass $(\hbar c/G)^{1/2}$	m_P	$2.1767(16) \times 10^{-8}$	kg	7.5×10^{-4}
Planck length $\hbar/m_{\text{PC}} = (\hbar G/c^3)^{1/2}$	l_P	$1.6160(12) \times 10^{-35}$	m	7.5×10^{-4}
Planck time $l_P/c = (\hbar G/c^5)^{1/2}$	t_P	$5.3906(40) \times 10^{-44}$	s	7.5×10^{-4}
ELECTROMAGNETIC				
elementary charge	e	$1.602\ 176\ 462(63) \times 10^{-19}$	C	3.9×10^{-8}
	e/h	$2.417\ 989\ 491(95) \times 10^{14}$	A J^{-1}	3.9×10^{-8}
magnetic flux quantum $h/2e$	Φ_0	$2.067\ 833\ 636(81) \times 10^{-15}$	Wb	3.9×10^{-8}
conductance quantum $2e^2/h$	G_0	$7.748\ 091\ 696(28) \times 10^{-5}$	S	3.7×10^{-9}
inverse of conductance quantum	G_0^{-1}	12 906.403 786(47)	Ω	3.7×10^{-9}
Josephson constant ^a $2e/h$	K_J	$483\ 597.898(19) \times 10^9$	Hz V^{-1}	3.9×10^{-8}
von Klitzing constant ^b $h/e^2 = \mu_0 c/2\alpha$	R_K	25 812.807 572(95)	Ω	3.7×10^{-9}
Bohr magneton $e\hbar/2m_e$ in eV T ⁻¹	μ_B	$927\ 400\ 899(37) \times 10^{-26}$ $5.788\ 381\ 749(43) \times 10^{-5}$	J T^{-1} eV T^{-1}	4.0×10^{-8} 7.3×10^{-9}
	μ_B/h	$13.996\ 246\ 24(56) \times 10^9$	Hz T ⁻¹	4.0×10^{-8}
	μ_B/hc	46.686 4521(19)	$\text{m}^{-1} \text{ T}^{-1}$	4.0×10^{-8}
	μ_B/k	0.671 7131(12)	K T ⁻¹	1.7×10^{-6}
nuclear magneton $e\hbar/2m_p$ in eV T ⁻¹	μ_N	$5.050\ 783\ 17(20) \times 10^{-27}$ $3.152\ 451\ 238(24) \times 10^{-8}$	J T^{-1} eV T^{-1}	4.0×10^{-8} 7.6×10^{-9}
	μ_N/h	7.622 593 96(31)	MHz T ⁻¹	4.0×10^{-8}
	μ_N/hc	$2.542\ 623\ 66(10) \times 10^{-2}$	$\text{m}^{-1} \text{ T}^{-1}$	4.0×10^{-8}
	μ_N/k	$3.658\ 2638(64) \times 10^{-4}$	K T ⁻¹	1.7×10^{-6}
ATOMIC AND NUCLEAR				
General				
fine-structure constant $e^2/4\pi\epsilon_0\hbar c$	α	$7.297\ 352\ 533(27) \times 10^{-3}$		3.7×10^{-9}
inverse fine-structure constant	α^{-1}	137.035 999 76(50)		3.7×10^{-9}

Fundamental Physical Constants

Quantity	Symbol	Value	Unit	Relative std. uncert. u_r
Rydberg constant $\alpha^2 m_e c / 2h$	R_∞	10 973 731.568 549(83)	m^{-1}	7.6×10^{-12}
	$R_\infty c$	3.289 841 960 368(25) $\times 10^{15}$	Hz	7.6×10^{-12}
	$R_\infty hc$	2.179 871 90(17) $\times 10^{-18}$	J	7.8×10^{-8}
$R_\infty hc$ in eV		13.605 691 72(53)	eV	3.9×10^{-8}
Bohr radius $\alpha / 4\pi R_\infty = 4\pi \epsilon_0 \hbar^2 / m_e e^2$	a_0	0.529 177 2083(19) $\times 10^{-10}$	m	3.7×10^{-9}
Hartree energy $e^2 / 4\pi \epsilon_0 a_0 = 2R_\infty hc$ $= \alpha^2 m_e c^2$ in eV	E_h	4.359 743 81(34) $\times 10^{-18}$	J	7.8×10^{-8}
quantum of circulation	$h/2m_e$	27.211 3834(11)	eV	3.9×10^{-8}
	h/m_e	3.636 947 516(27) $\times 10^{-4}$	$\text{m}^2 \text{s}^{-1}$	7.3×10^{-9}
		7.273 895 032(53) $\times 10^{-4}$	$\text{m}^2 \text{s}^{-1}$	7.3×10^{-9}
Electroweak				
Fermi coupling constant ^c	$G_F / (\hbar c)^3$	1.166 39(1) $\times 10^{-5}$	GeV^{-2}	8.6×10^{-6}
weak mixing angle ^d θ_W (on-shell scheme) $\sin^2 \theta_W = s_W^2 \equiv 1 - (m_W/m_Z)^2$	$\sin^2 \theta_W$	0.2224(19)		8.7×10^{-3}
Electron, e^-				
electron mass in u, $m_e = A_r(e) u$ (electron relative atomic mass times u) energy equivalent in MeV	m_e	9.109 381 88(72) $\times 10^{-31}$	kg	7.9×10^{-8}
	$m_e c^2$	5.485 799 110(12) $\times 10^{-4}$	u	2.1×10^{-9}
		8.187 104 14(64) $\times 10^{-14}$	J	7.9×10^{-8}
		0.510 998 902(21)	MeV	4.0×10^{-8}
electron-muon mass ratio	m_e/m_μ	4.836 332 10(15) $\times 10^{-3}$		3.0×10^{-8}
electron-tau mass ratio	m_e/m_τ	2.875 55(47) $\times 10^{-4}$		1.6×10^{-4}
electron-proton mass ratio	m_e/m_p	5.446 170 232(12) $\times 10^{-4}$		2.1×10^{-9}
electron-neutron mass ratio	m_e/m_n	5.438 673 462(12) $\times 10^{-4}$		2.2×10^{-9}
electron-deuteron mass ratio	m_e/m_d	2.724 437 1170(58) $\times 10^{-4}$		2.1×10^{-9}
electron to alpha particle mass ratio	m_e/m_α	1.370 933 5611(29) $\times 10^{-4}$		2.1×10^{-9}
electron charge to mass quotient	$-e/m_e$	-1.758 820 174(71) $\times 10^{11}$	C kg^{-1}	4.0×10^{-8}
electron molar mass $N_A m_e$	$M(e), M_e$	5.485 799 110(12) $\times 10^{-7}$	kg mol^{-1}	2.1×10^{-9}
Compton wavelength $h/m_e c$ $\lambda_C / 2\pi = \alpha a_0 = \alpha^2 / 4\pi R_\infty$	λ_C	2.426 310 215(18) $\times 10^{-12}$	m	7.3×10^{-9}
classical electron radius $\alpha^2 a_0$	λ_C	386.159 2642(28) $\times 10^{-15}$	m	7.3×10^{-9}
Thomson cross section $(8\pi/3)r_e^2$	r_e	2.817 940 285(31) $\times 10^{-15}$	m	1.1×10^{-8}
	σ_e	0.665 245 854(15) $\times 10^{-28}$	m^2	2.2×10^{-8}
electron magnetic moment to Bohr magneton ratio to nuclear magneton ratio	μ_e	-928.476 362(37) $\times 10^{-26}$	J T^{-1}	4.0×10^{-8}
electron magnetic moment anomaly $ \mu_e /\mu_B - 1$	μ_e/μ_B	-1.001 159 652 1869(41)		4.1×10^{-12}
electron g-factor $-2(1 + a_e)$	μ_e/μ_N	-1 838.281 9660(39)		2.1×10^{-9}
	a_e	1.159 652 1869(41) $\times 10^{-3}$		3.5×10^{-9}
	g_e	-2.002 319 304 3737(82)		4.1×10^{-12}
electron-muon magnetic moment ratio	μ_e/μ_μ	206.766 9720(63)		3.0×10^{-8}

Fundamental Physical Constants

Quantity	Symbol	Value	Unit	Relative std. uncert. u_r
electron-proton magnetic moment ratio	μ_e/μ_p	- 658.210 6875(66)		1.0×10^{-8}
electron to shielded proton magnetic moment ratio (H ₂ O, sphere, 25 °C)	μ_e/μ'_p	- 658.227 5954(71)		1.1×10^{-8}
electron-neutron magnetic moment ratio	μ_e/μ_n	960.920 50(23)		2.4×10^{-7}
electron-deuteron magnetic moment ratio	μ_e/μ_d	- 2 143.923 498(23)		1.1×10^{-8}
electron to shielded helion ^e magnetic moment ratio (gas, sphere, 25 °C)	μ_e/μ'_h	864.058 255(10)		1.2×10^{-8}
electron gyromagnetic ratio $2 \mu_e /\hbar$	γ_e	1.760 859 794(71) $\times 10^{11}$	s ⁻¹ T ⁻¹	4.0×10^{-8}
	$\gamma_e/2\pi$	28 024.9540(11)	MHz T ⁻¹	4.0×10^{-8}
Muon, μ^-				
muon mass in u, $m_\mu = A_r(\mu)$ u (muon)	m_μ	1.883 531 09(16) $\times 10^{-28}$	kg	8.4×10^{-8}
relative atomic mass times u		0.113 428 9168(34)	u	3.0×10^{-8}
energy equivalent in MeV	$m_\mu c^2$	1.692 833 32(14) $\times 10^{-11}$	J	8.4×10^{-8}
		105.658 3568(52)	MeV	4.9×10^{-8}
muon-electron mass ratio	m_μ/m_e	206.768 2657(63)		3.0×10^{-8}
muon-tau mass ratio	m_μ/m_τ	5.945 72(97) $\times 10^{-2}$		1.6×10^{-4}
muon-proton mass ratio	m_μ/m_p	0.112 609 5173(34)		3.0×10^{-8}
muon-neutron mass ratio	m_μ/m_n	0.112 454 5079(34)		3.0×10^{-8}
muon molar mass $N_A m_\mu$	$M(\mu), M_\mu$	0.113 428 9168(34) $\times 10^{-3}$	kg mol ⁻¹	3.0×10^{-8}
muon Compton wavelength $h/m_\mu c$ $\lambda_{C,\mu}/2\pi$	$\lambda_{C,\mu}$	11.734 441 97(35) $\times 10^{-15}$	m	2.9×10^{-8}
muon magnetic moment to Bohr magneton ratio	μ_μ	1.867 594 444(55) $\times 10^{-15}$	m	2.9×10^{-8}
to nuclear magneton ratio	μ_μ/μ_B	- 4.490 448 13(22) $\times 10^{-26}$	J T ⁻¹	4.9×10^{-8}
	μ_μ/μ_N	- 4.841 970 85(15) $\times 10^{-3}$		3.0×10^{-8}
		- 8.890 597 70(27)		3.0×10^{-8}
muon magnetic moment anomaly $ \mu_\mu /(e\hbar/2m_\mu) - 1$	a_μ	1.165 916 02(64) $\times 10^{-3}$		5.5×10^{-7}
muon g-factor $-2(1 + a_\mu)$	g_μ	- 2.002 331 8320(13)		6.4×10^{-10}
muon-proton magnetic moment ratio	μ_μ/μ_p	- 3.183 345 39(10)		3.2×10^{-8}
Tau, τ^-				
tau mass ^f in u, $m_\tau = A_r(\tau)$ u (tau)	m_τ	3.167 88(52) $\times 10^{-27}$	kg	1.6×10^{-4}
relative atomic mass times u		1.907 74(31)	u	1.6×10^{-4}
energy equivalent in MeV	$m_\tau c^2$	2.847 15(46) $\times 10^{-10}$	J	1.6×10^{-4}
		1 777.05(29)	MeV	1.6×10^{-4}

Fundamental Physical Constants

Quantity	Symbol	Value	Unit	Relative std. uncert. u_r
tau-electron mass ratio	m_τ/m_e	3 477.60(57)		1.6×10^{-4}
tau-muon mass ratio	m_τ/m_μ	16.8188(27)		1.6×10^{-4}
tau-proton mass ratio	m_τ/m_p	1.893 96(31)		1.6×10^{-4}
tau-neutron mass ratio	m_τ/m_n	1.891 35(31)		1.6×10^{-4}
tau molar mass $N_A m_\tau$	$M(\tau), M_\tau$	$1.907\,74(31) \times 10^{-3}$	kg mol^{-1}	1.6×10^{-4}
tau Compton wavelength $h/m_\tau c$	$\lambda_{C,\tau}$	$0.697\,70(11) \times 10^{-15}$	m	1.6×10^{-4}
$\lambda_{C,\tau}/2\pi$	$\tilde{\lambda}_{C,\tau}$	$0.111\,042(18) \times 10^{-15}$	m	1.6×10^{-4}
Proton, p				
proton mass in u, $m_p = A_r(p) u$ (proton relative atomic mass times u)	m_p	$1.672\,621\,58(13) \times 10^{-27}$	kg	7.9×10^{-8}
energy equivalent in MeV	$m_p c^2$	1.007 276 466 88(13) $1.503\,277\,31(12) \times 10^{-10}$ 938.271 998(38)	u J MeV	1.3×10^{-10} 7.9×10^{-8} 4.0×10^{-8}
proton-electron mass ratio	m_p/m_e	1 836.152 6675(39)		2.1×10^{-9}
proton-muon mass ratio	m_p/m_μ	8.880 244 08(27)		3.0×10^{-8}
proton-tau mass ratio	m_p/m_τ	0.527 994(86)		1.6×10^{-4}
proton-neutron mass ratio	m_p/m_n	0.998 623 478 55(58)		5.8×10^{-10}
proton charge to mass quotient	e/m_p	$9.578\,834\,08(38) \times 10^7$	C kg^{-1}	4.0×10^{-8}
proton molar mass $N_A m_p$	$M(p), M_p$	$1.007\,276\,466\,88(13) \times 10^{-3}$	kg mol^{-1}	1.3×10^{-10}
proton Compton wavelength $h/m_p c$	$\lambda_{C,p}$	$1.321\,409\,847(10) \times 10^{-15}$	m	7.6×10^{-9}
$\lambda_{C,p}/2\pi$	$\tilde{\lambda}_{C,p}$	$0.210\,308\,9089(16) \times 10^{-15}$	m	7.6×10^{-9}
proton magnetic moment to Bohr magneton ratio	μ_p	$1.410\,606\,633(58) \times 10^{-26}$	J T^{-1}	4.1×10^{-8}
to nuclear magneton ratio	μ_p/μ_N	$1.521\,032\,203(15) \times 10^{-3}$		1.0×10^{-8}
proton g-factor $2\mu_p/\mu_N$	g_p	2.792 847 337(29)		1.0×10^{-8}
proton-neutron magnetic moment ratio	μ_p/μ_n	-1.459 898 05(34)		2.4×10^{-7}
shielded proton magnetic moment (H ₂ O, sphere, 25 °C)	μ'_p	$1.410\,570\,399(59) \times 10^{-26}$	J T^{-1}	4.2×10^{-8}
to Bohr magneton ratio	μ'_p/μ_B	$1.520\,993\,132(16) \times 10^{-3}$		1.1×10^{-8}
to nuclear magneton ratio	μ'_p/μ_N	2.792 775 597(31)		1.1×10^{-8}
proton magnetic shielding correction $1 - \mu'_p/\mu_p$ (H ₂ O, sphere, 25 °C)	σ'_p	$25.687(15) \times 10^{-6}$		5.7×10^{-4}
proton gyromagnetic ratio $2\mu_p/\hbar$	γ_p	$2.675\,222\,12(11) \times 10^8$	$\text{s}^{-1} \text{T}^{-1}$	4.1×10^{-8}
	$\gamma_p/2\pi$	42.577 4825(18)	MHz T^{-1}	4.1×10^{-8}
shielded proton gyromagnetic ratio $2\mu'_p/\hbar$ (H ₂ O, sphere, 25 °C)	γ'_p	$2.675\,153\,41(11) \times 10^8$	$\text{s}^{-1} \text{T}^{-1}$	4.2×10^{-8}
	$\gamma'_p/2\pi$	42.576 3888(18)	MHz T^{-1}	4.2×10^{-8}
Neutron, n				

Fundamental Physical Constants

Quantity	Symbol	Value	Unit	Relative std. uncert. u_r
neutron mass in u, $m_n = A_r(n) u$ (neutron relative atomic mass times u)	m_n	$1.674\,927\,16(13) \times 10^{-27}$	kg	7.9×10^{-8}
energy equivalent in MeV	$m_n c^2$	$1.008\,664\,915\,78(55)$ $1.505\,349\,46(12) \times 10^{-10}$ 939.565 330(38)	u J MeV	5.4×10^{-10} 7.9×10^{-8} 4.0×10^{-8}
neutron-electron mass ratio	m_n/m_e	1 838.683 6550(40)		2.2×10^{-9}
neutron-muon mass ratio	m_n/m_μ	8.892 484 78(27)		3.0×10^{-8}
neutron-tau mass ratio	m_n/m_τ	0.528 722(86)		1.6×10^{-4}
neutron-proton mass ratio	m_n/m_p	1.001 378 418 87(58)		5.8×10^{-10}
neutron molar mass $N_A m_n$	$M(n), M_n$	$1.008\,664\,915\,78(55) \times 10^{-3}$	kg mol ⁻¹	5.4×10^{-10}
neutron Compton wavelength $h/m_n c$ $\lambda_{C,n}/2\pi$	$\lambda_{C,n}$	$1.319\,590\,898(10) \times 10^{-15}$	m	7.6×10^{-9}
neutron magnetic moment to Bohr magneton ratio	μ_n	$0.210\,019\,4142(16) \times 10^{-15}$	m	7.6×10^{-9}
to nuclear magneton ratio	μ_n/μ_B	$-0.966\,236\,40(23) \times 10^{-26}$	J T ⁻¹	2.4×10^{-7}
	μ_n/μ_N	$-1.041\,875\,63(25) \times 10^{-3}$		2.4×10^{-7}
		$-1.913\,042\,72(45)$		2.4×10^{-7}
neutron g-factor $2\mu_n/\mu_N$	g_n	-3.826 085 45(90)		2.4×10^{-7}
neutron-electron magnetic moment ratio	μ_n/μ_e	$1.040\,668\,82(25) \times 10^{-3}$		2.4×10^{-7}
neutron-proton magnetic moment ratio	μ_n/μ_p	-0.684 979 34(16)		2.4×10^{-7}
neutron to shielded proton magnetic moment ratio (H ₂ O, sphere, 25 °C)	μ_n/μ'_p	-0.684 996 94(16)		2.4×10^{-7}
neutron gyromagnetic ratio $2 \mu_n /\hbar$	γ_n	$1.832\,471\,88(44) \times 10^8$	s ⁻¹ T ⁻¹	2.4×10^{-7}
	$\gamma_n/2\pi$	29.164 6958(70)	MHz T ⁻¹	2.4×10^{-7}
Deuteron, d				
deuteron mass in u, $m_d = A_r(d) u$ (deuteron relative atomic mass times u)	m_d	$3.343\,583\,09(26) \times 10^{-27}$	kg	7.9×10^{-8}
energy equivalent in MeV	$m_d c^2$	$2.013\,553\,212\,71(35)$ $3.005\,062\,62(24) \times 10^{-10}$ 1 875.612 762(75)	u J MeV	1.7×10^{-10} 7.9×10^{-8} 4.0×10^{-8}
deuteron-electron mass ratio	m_d/m_e	3 670.482 9550(78)		2.1×10^{-9}
deuteron-proton mass ratio	m_d/m_p	1.999 007 500 83(41)		2.0×10^{-10}
deuteron molar mass $N_A m_d$	$M(d), M_d$	$2.013\,553\,212\,71(35) \times 10^{-3}$	kg mol ⁻¹	1.7×10^{-10}
deuteron magnetic moment to Bohr magneton ratio	μ_d	$0.433\,073\,457(18) \times 10^{-26}$	J T ⁻¹	4.2×10^{-8}
to nuclear magneton ratio	μ_d/μ_B	$0.466\,975\,4556(50) \times 10^{-3}$		1.1×10^{-8}
	μ_d/μ_N	0.857 438 2284(94)		1.1×10^{-8}
deuteron-electron magnetic moment ratio	μ_d/μ_e	$-4.664\,345\,537(50) \times 10^{-4}$		1.1×10^{-8}
deuteron-proton magnetic moment ratio	μ_d/μ_p	0.307 012 2083(45)		1.5×10^{-8}

Fundamental Physical Constants

Quantity	Symbol	Value	Unit	Relative std. uncert. u_r
deuteron-neutron magnetic moment ratio	μ_d/μ_n	-0.448 206 52(11)		2.4×10^{-7}
		Helion, h		
helion mass ^e in u, $m_h = A_r(h) u$ (helion relative atomic mass times u)	m_h	$5.006\,411\,74(39) \times 10^{-27}$	kg	7.9×10^{-8}
energy equivalent in MeV	$m_h c^2$	$3.014\,932\,234\,69(86)$ $4.499\,538\,48(35) \times 10^{-10}$ 2 808.391 32(11)	u J MeV	2.8×10^{-10} 7.9×10^{-8} 4.0×10^{-8}
helion-electron mass ratio	m_h/m_e	5 495.885 238(12)		2.1×10^{-9}
helion-proton mass ratio	m_h/m_p	2.993 152 658 50(93)		3.1×10^{-10}
helion molar mass $N_A m_h$	$M(h), M_h$	$3.014\,932\,234\,69(86) \times 10^{-3}$	kg mol ⁻¹	2.8×10^{-10}
shielded helion magnetic moment (gas, sphere, 25 °C)	μ'_h	$-1.074\,552\,967(45) \times 10^{-26}$	J T ⁻¹	4.2×10^{-8}
to Bohr magneton ratio	μ'_h/μ_B	$-1.158\,671\,474(14) \times 10^{-3}$		1.2×10^{-8}
to nuclear magneton ratio	μ'_h/μ_N	-2.127 497 718(25)		1.2×10^{-8}
shielded helion to proton magnetic moment ratio (gas, sphere, 25 °C)	μ'_h/μ_p	-0.761 766 563(12)		1.5×10^{-8}
shielded helion to shielded proton magnetic moment ratio (gas/H ₂ O, spheres, 25 °C)	μ'_h/μ'_p	-0.761 786 1313(33)		4.3×10^{-9}
shielded helion gyromagnetic ratio $2 \mu'_h /\hbar$ (gas, sphere, 25 °C)	γ'_h	$2.037\,894\,764(85) \times 10^8$	s ⁻¹ T ⁻¹	4.2×10^{-8}
	$\gamma'_h/2\pi$	32.434 1025(14)	MHz T ⁻¹	4.2×10^{-8}
		Alpha particle, α		
alpha particle mass in u, $m_\alpha = A_r(\alpha) u$ (alpha particle relative atomic mass times u)	m_α	$6.644\,655\,98(52) \times 10^{-27}$	kg	7.9×10^{-8}
energy equivalent in MeV	$m_\alpha c^2$	$4.001\,506\,1747(10)$ $5.971\,918\,97(47) \times 10^{-10}$ 3 727.379 04(15)	u J MeV	2.5×10^{-10} 7.9×10^{-8} 4.0×10^{-8}
alpha particle to electron mass ratio	m_α/m_e	7 294.299 508(16)		2.1×10^{-9}
alpha particle to proton mass ratio	m_α/m_p	3.972 599 6846(11)		2.8×10^{-10}
alpha particle molar mass $N_A m_\alpha$	$M(\alpha), M_\alpha$	$4.001\,506\,1747(10) \times 10^{-3}$	kg mol ⁻¹	2.5×10^{-10}
		PHYSICO-CHEMICAL		
Avogadro constant	N_A, L	$6.022\,141\,99(47) \times 10^{23}$	mol ⁻¹	7.9×10^{-8}
atomic mass constant	m_u	$1.660\,538\,73(13) \times 10^{-27}$	kg	7.9×10^{-8}
$m_u = \frac{1}{12}m(^{12}\text{C}) = 1 \text{ u}$ $= 10^{-3} \text{ kg mol}^{-1}/N_A$ energy equivalent in MeV	$m_u c^2$	$1.492\,417\,78(12) \times 10^{-10}$ 931.494 013(37)	J MeV	7.9×10^{-8} 4.0×10^{-8}
Faraday constant ^g $N_A e$	F	96 485.3415(39)	C mol ⁻¹	4.0×10^{-8}

Fundamental Physical Constants

Quantity	Symbol	Value	Unit	Relative std. uncert. u_r
molar Planck constant	$N_A h$	$3.990\,312\,689(30) \times 10^{-10}$	J s mol ⁻¹	7.6×10^{-9}
	$N_A h c$	0.119 626 564 92(91)	J m mol ⁻¹	7.6×10^{-9}
molar gas constant	R	8.314 472(15)	J mol ⁻¹ K ⁻¹	1.7×10^{-6}
Boltzmann constant R/N_A in eV K ⁻¹	k	$1.380\,6503(24) \times 10^{-23}$	J K ⁻¹	1.7×10^{-6}
		8.617 342(15) $\times 10^{-5}$	eV K ⁻¹	1.7×10^{-6}
	k/h	$2.083\,6644(36) \times 10^{10}$	Hz K ⁻¹	1.7×10^{-6}
	k/hc	69.503 56(12)	m ⁻¹ K ⁻¹	1.7×10^{-6}
molar volume of ideal gas RT/p $T = 273.15$ K, $p = 101.325$ kPa	V_m	$22.413\,996(39) \times 10^{-3}$	m ³ mol ⁻¹	1.7×10^{-6}
Loschmidt constant N_A/V_m	n_0	2.686 7775(47) $\times 10^{25}$	m ⁻³	1.7×10^{-6}
$T = 273.15$ K, $p = 100$ kPa	V_m	$22.710\,981(40) \times 10^{-3}$	m ³ mol ⁻¹	1.7×10^{-6}
Sackur-Tetrode constant (absolute entropy constant) ^h $\frac{5}{2} + \ln[(2\pi m_u k T_1/h^2)^{3/2} k T_1/p_0]$ $T_1 = 1$ K, $p_0 = 100$ kPa	S_0/R	-1.151 7048(44)		3.8×10^{-6}
$T_1 = 1$ K, $p_0 = 101.325$ kPa		-1.164 8678(44)		3.7×10^{-6}
Stefan-Boltzmann constant $(\pi^2/60)k^4/\hbar^3c^2$	σ	$5.670\,400(40) \times 10^{-8}$	W m ⁻² K ⁻⁴	7.0×10^{-6}
first radiation constant $2\pi hc^2$	c_1	$3.741\,771\,07(29) \times 10^{-16}$	W m ²	7.8×10^{-8}
first radiation constant for spectral radiance $2hc^2$	c_{1L}	$1.191\,042\,722(93) \times 10^{-16}$	W m ² sr ⁻¹	7.8×10^{-8}
second radiation constant hc/k	c_2	$1.438\,7752(25) \times 10^{-2}$	m K	1.7×10^{-6}
Wien displacement law constant $b = \lambda_{\max} T = c_2/4.965\,114\,231\dots$	b	$2.897\,7686(51) \times 10^{-3}$	m K	1.7×10^{-6}

^a See the “Adopted values” table for the conventional value adopted internationally for realizing representations of the volt using the Josephson effect.

^b See the “Adopted values” table for the conventional value adopted internationally for realizing representations of the ohm using the quantum Hall effect.

^c Value recommended by the Particle Data Group, Caso et al., Eur. Phys. J. C **3**(1-4), 1-794 (1998).

^d Based on the ratio of the masses of the W and Z bosons m_W/m_Z recommended by the Particle Data Group (Caso et al., 1998). The value for $\sin^2\theta_W$ they recommend, which is based on a particular variant of the modified minimal subtraction ($\overline{\text{MS}}$) scheme, is $\sin^2\hat{\theta}_W(M_Z) = 0.231\,24(24)$.

^e The helion, symbol h, is the nucleus of the ^3He atom.

^f This and all other values involving m_τ are based on the value of $m_\tau c^2$ in MeV recommended by the Particle Data Group, Caso et al., Eur. Phys. J. C **3**(1-4), 1-794 (1998), but with a standard uncertainty of 0.29 MeV rather than the quoted uncertainty of -0.26 MeV, +0.29 MeV.

^g The numerical value of F to be used in coulometric chemical measurements is 96 485.3432(76) [7.9×10^{-8}] when the relevant current is measured in terms of representations of the volt and ohm based on the Josephson and quantum Hall effects and the internationally adopted conventional values of the Josephson and von Klitzing constants K_J -90 and R_K -90 given in the “Adopted values” table.

^h The entropy of an ideal monoatomic gas of relative atomic mass A_r is given by $S = S_0 + \frac{3}{2}R \ln A_r - R \ln(p/p_0) + \frac{5}{2}R \ln(T/K)$.

Fundamental Physical Constants — Adopted values

Quantity	Symbol	Value	Unit	Relative std. uncert. u_r
molar mass of ^{12}C	$M(^{12}\text{C})$	12×10^{-3}	kg mol^{-1}	(exact)
molar mass constant ^a $M(^{12}\text{C})/12$	M_u	1×10^{-3}	kg mol^{-1}	(exact)
conventional value of Josephson constant ^b	$K_{\text{J}-90}$	483 597.9	GHz V^{-1}	(exact)
conventional value of von Klitzing constant ^c	$R_{\text{K}-90}$	25 812.807	Ω	(exact)
standard atmosphere		101 325	Pa	(exact)
standard acceleration of gravity	g_n	9.806 65	m s^{-2}	(exact)

^a The relative atomic mass $A_r(\text{X})$ of particle X with mass $m(\text{X})$ is defined by $A_r(\text{X}) = m(\text{X})/m_u$, where $m_u = m(^{12}\text{C})/12 = M_u/N_A = 1 \text{ u}$ is the atomic mass constant, N_A is the Avogadro constant, and u is the atomic mass unit. Thus the mass of particle X in u is $m(\text{X}) = A_r(\text{X}) u$ and the molar mass of X is $M(\text{X}) = A_r(\text{X}) M_u$.

^b This is the value adopted internationally for realizing representations of the volt using the Josephson effect.

^c This is the value adopted internationally for realizing representations of the ohm using the quantum Hall effect.

Energy Equivalents

	J	kg	m^{-1}	Hz
1 J	$(1 \text{ J}) =$ 1 J	$(1 \text{ J})/c^2 =$ $1.112\,650\,056 \times 10^{-17} \text{ kg}$	$(1 \text{ J})/hc =$ $5.034\,117\,62(39) \times 10^{24} \text{ m}^{-1}$	$(1 \text{ J})/h =$ $1.509\,190\,50(12) \times 10^{33} \text{ Hz}$
1 kg	$(1 \text{ kg})c^2 =$ $8.987\,551\,787 \times 10^{16} \text{ J}$	$(1 \text{ kg}) =$ 1 kg	$(1 \text{ kg})c/h =$ $4.524\,439\,29(35) \times 10^{41} \text{ m}^{-1}$	$(1 \text{ kg})c^2/h =$ $1.356\,392\,77(11) \times 10^{50} \text{ Hz}$
1 m^{-1}	$(1 \text{ m}^{-1})hc =$ $1.986\,445\,44(16) \times 10^{-25} \text{ J}$	$(1 \text{ m}^{-1})h/c =$ $2.210\,218\,63(17) \times 10^{-42} \text{ kg}$	$(1 \text{ m}^{-1}) =$ 1 m^{-1}	$(1 \text{ m}^{-1})c =$ $299\,792\,458 \text{ Hz}$
1 Hz	$(1 \text{ Hz})h =$ $6.626\,068\,76(52) \times 10^{-34} \text{ J}$	$(1 \text{ Hz})h/c^2 =$ $7.372\,495\,78(58) \times 10^{-51} \text{ kg}$	$(1 \text{ Hz})c =$ $3.335\,640\,952 \times 10^{-9} \text{ m}^{-1}$	$(1 \text{ Hz}) =$ 1 Hz
1 K	$(1 \text{ K})k =$ $1.380\,6503(24) \times 10^{-23} \text{ J}$	$(1 \text{ K})k/c^2 =$ $1.536\,1807(27) \times 10^{-40} \text{ kg}$	$(1 \text{ K})k/hc =$ $69.503\,56(12) \text{ m}^{-1}$	$(1 \text{ K})k/h =$ $2.083\,6644(36) \times 10^{10} \text{ Hz}$
1 eV	$(1 \text{ eV}) =$ $1.602\,176\,462(63) \times 10^{-19} \text{ J}$	$(1 \text{ eV})/c^2 =$ $1.782\,661\,731(70) \times 10^{-36} \text{ kg}$	$(1 \text{ eV})/hc =$ $8.065\,544\,77(32) \times 10^5 \text{ m}^{-1}$	$(1 \text{ eV})/h =$ $2.417\,989\,491(95) \times 10^{14} \text{ Hz}$
1 u	$(1 \text{ u})c^2 =$ $1.492\,417\,78(12) \times 10^{-10} \text{ J}$	$(1 \text{ u}) =$ $1.660\,538\,73(13) \times 10^{-27} \text{ kg}$	$(1 \text{ u})c/h =$ $7.513\,006\,658(57) \times 10^{14} \text{ m}^{-1}$	$(1 \text{ u})c^2/h =$ $2.252\,342\,733(17) \times 10^{23} \text{ Hz}$
$1 E_{\text{h}}$	$(1 E_{\text{h}}) =$ $4.359\,743\,81(34) \times 10^{-18} \text{ J}$	$(1 E_{\text{h}})/c^2 =$ $4.850\,869\,19(38) \times 10^{-35} \text{ kg}$	$(1 E_{\text{h}})/hc =$ $2.194\,746\,313\,710(17) \times 10^7 \text{ m}^{-1}$	$(1 E_{\text{h}})/h =$ $6.579\,683\,920\,735(50) \times 10^{15} \text{ Hz}$

Derived from the relations $E = mc^2 = hc/\lambda = h\nu = kT$, and based on the 1998 CODATA adjustment of the values of the constants;

$1 \text{ eV} = (e/C) \text{ J}$, $1 \text{ u} = m_{\text{u}} = \frac{1}{12}m(^{12}\text{C}) = 10^{-3} \text{ kg mol}^{-1}/N_A$, and $E_{\text{h}} = 2R_{\infty}hc = \alpha^2 m_{\text{e}}c^2$ is the Hartree energy (hartree).

Energy Equivalents

	K	eV	u	E_h
1 J	$(1 \text{ J})/k = 7.242\,964(13) \times 10^{22} \text{ K}$	$(1 \text{ J}) = 6.241\,509\,74(24) \times 10^{18} \text{ eV}$	$(1 \text{ J})/c^2 = 6.700\,536\,62(53) \times 10^9 \text{ u}$	$(1 \text{ J}) = 2.293\,712\,76(18) \times 10^{17} E_h$
1 kg	$(1 \text{ kg})c^2/k = 6.509\,651(11) \times 10^{39} \text{ K}$	$(1 \text{ kg})c^2 = 5.609\,589\,21(22) \times 10^{35} \text{ eV}$	$(1 \text{ kg}) = 6.022\,141\,99(47) \times 10^{26} \text{ u}$	$(1 \text{ kg})c^2 = 2.061\,486\,22(16) \times 10^{34} E_h$
1 m^{-1}	$(1 \text{ m}^{-1})hc/k = 1.438\,7752(25) \times 10^{-2} \text{ K}$	$(1 \text{ m}^{-1})hc = 1.239\,841\,857(49) \times 10^{-6} \text{ eV}$	$(1 \text{ m}^{-1})h/c = 1.331\,025\,042(10) \times 10^{-15} \text{ u}$	$(1 \text{ m}^{-1})hc = 4.556\,335\,252\,750(35) \times 10^{-8} E_h$
1 Hz	$(1 \text{ Hz})h/k = 4.799\,2374(84) \times 10^{-11} \text{ K}$	$(1 \text{ Hz})h = 4.135\,667\,27(16) \times 10^{-15} \text{ eV}$	$(1 \text{ Hz})h/c^2 = 4.439\,821\,637(34) \times 10^{-24} \text{ u}$	$(1 \text{ Hz})h = 1.519\,829\,846\,003(12) \times 10^{-16} E_h$
1 K	$(1 \text{ K}) = 1 \text{ K}$	$(1 \text{ K})k = 8.617\,342(15) \times 10^{-5} \text{ eV}$	$(1 \text{ K})k/c^2 = 9.251\,098(16) \times 10^{-14} \text{ u}$	$(1 \text{ K})k = 3.166\,8153(55) \times 10^{-6} E_h$
1 eV	$(1 \text{ eV})/k = 1.160\,4506(20) \times 10^4 \text{ K}$	$(1 \text{ eV}) = 1 \text{ eV}$	$(1 \text{ eV})/c^2 = 1.073\,544\,206(43) \times 10^{-9} \text{ u}$	$(1 \text{ eV}) = 3.674\,932\,60(14) \times 10^{-2} E_h$
1 u	$(1 \text{ u})c^2/k = 1.080\,9528(19) \times 10^{13} \text{ K}$	$(1 \text{ u})c^2 = 931.494\,013(37) \times 10^6 \text{ eV}$	$(1 \text{ u}) = 1 \text{ u}$	$(1 \text{ u})c^2 = 3.423\,177\,709(26) \times 10^7 E_h$
$1 E_h$	$(1 E_h)/k = 3.157\,7465(55) \times 10^5 \text{ K}$	$(1 E_h) = 27.211\,3834(11) \text{ eV}$	$(1 E_h)/c^2 = 2.921\,262\,304(22) \times 10^{-8} \text{ u}$	$(1 E_h) = 1 E_h$

Derived from the relations $E = mc^2 = hc/\lambda = h\nu = kT$, and based on the 1998 CODATA adjustment of the values of the constants;
 $1 \text{ eV} = (e/C) \text{ J}$, $1 \text{ u} = m_u = \frac{1}{12}m^{12}\text{C} = 10^{-3} \text{ kg mol}^{-1}/N_A$, and $E_h = 2R_\infty hc = \alpha^2 m_e c^2$ is the Hartree energy (hartree).